

Analysis of Transition Metal Complex Excited States: Turning Numbers into Chemical Insight

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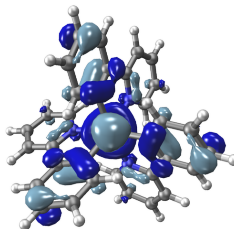


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Introduction

Computational Photochemistry

- ▶ Accurate numbers
- 😊 *Quantum chemical methods*:
Semi-emp., TDDFT, CC, ADC, CASSCF, DMRG, CASPT2, MR-CI, ...
- 😊 *Environmental models*: QM/MM, PCM, density embedding, ...
- 😊 *Algorithmic efforts*: Linear scaling, density fitting, parallelization, GPUs, ...
- ▶ Comparison to experiment
- 😊 *Linear* and *non-linear* optical properties
- 😊 *Static* and *time-resolved* experiments
- ▶ Chemical insight
- 😞 **Look at the HOMO and LUMO**

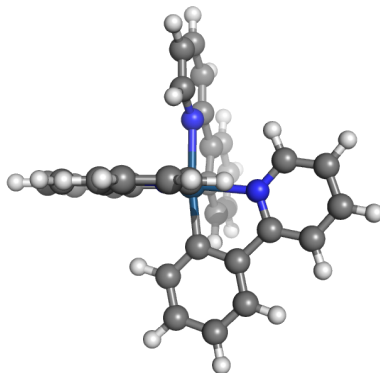


Ir(ppy)₃

Ir(ppy)₃

- ▶ Highly phosphorescent complex¹
- ▶ **Interplay** of LC and MLCT states decisive for fluorescence²
 - **MLCT** → relativistic effects → spin-orbit coupling
 - **IL** → low-energy triplets

⑦ How to quantify

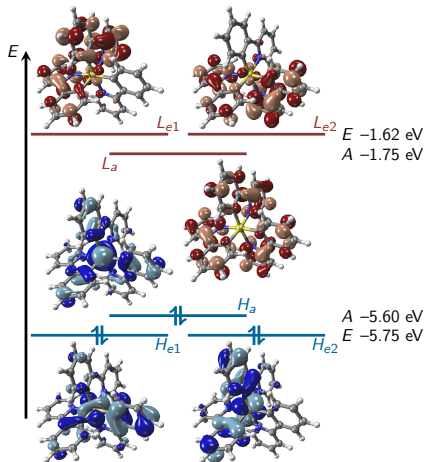


¹K. Dedeian, P. I. Djurovich, et al., *Inorg. Chem.* **1991**, 30, 1685.

²B. Powell, *CCR* **2015**, 295, 46.

Frontier orbitals

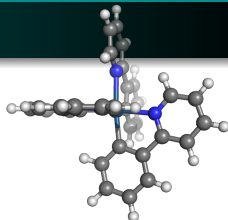
- ▶ DFT/B3LYP
- ▶ **Occupied** orbitals
 - Mixture metal-d/ligand- π
- ▶ **Virtual** orbitals
 - Ligand- π^*



Frontier orbitals

Lowest triplet states

► TDDFT/B3LYP



	E (eV)	Sym.	Leading excitations		
T_1	2.89	A	$+0.85H_aL_a$	$+0.32H_{e1}L_{e2}$	$+0.32H_{e2}L_{e1}$
T_2	2.93	E	$+0.58H_aL_{e1}$	$+0.49H_aL_{e2}$	
T_3	2.93	E	$+0.58H_aL_{e2}$	$-0.49H_aL_{e1}$	
T_4	3.17	$A+E$	$+0.48H_{e1}L_{e1}$	$+0.44H_{e2}L_{e1}$	$+0.41H_{e1}L_a$
T_5	3.17	$A+E$	$+0.56H_{e2}L_a$	$+0.46H_{e2}L_{e2}$	$-0.42H_{e2}L_{e1}$
T_6	3.17	$A+E$	$-0.56H_{e1}L_a$	$+0.53H_{e1}L_{e2}$	$+0.45H_aL_{e2}$

► Metal-d/ligand- $\pi \rightarrow$ ligand- π^* excitations

→ Mixture: MLCT, IL, LLCT

☹ Tedious **work**, possible **ambiguities**

❓ Different graphical representation

Natural Transition Orbitals

1-electron transition density matrix

$$D_{\mu\nu}^{0I} = \langle \Psi_0 | \hat{a}_\mu^\dagger \hat{a}_\nu | \Psi_I \rangle$$

$\hat{a}_\mu^\dagger$ Creation operator

$\hat{a}_\nu$ Annihilation operator

$D_{\mu\nu}^{0I}$ Matrix representation of the 1TDM

Natural transition orbitals — Singular value decomposition

$$\mathbf{D}^{0I} = \mathbf{U} \times \text{diag}(\sqrt{\lambda_1}, \sqrt{\lambda_2}, \dots) \times \mathbf{V}^T$$

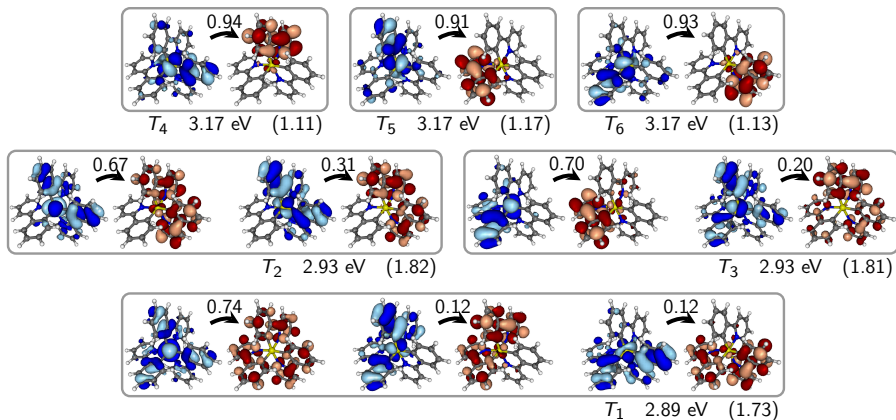
U **Hole** orbital coefficients

λ_i Transition amplitudes

V **Electron** orbital coefficients

¹R. L. Martin, *J. Chem. Phys.* **2003**, 11, 4775.

Natural transition orbitals



► Clear differences visible

→ More **IL** for $T_1 - T_3$, more **LLCT** for $T_4 - T_6$

⑦ Quantitative description

Charge Transfer Numbers

- Summation over **squared 1TDM elements**

Charge transfer numbers

$$\Omega_{AB} = \frac{1}{2} \sum_{\mu \in A} \sum_{\nu \in B} |\tilde{D}_{\mu\nu}^{0I}|^2$$

$\tilde{D}_{\mu\nu}^{0I}$ Element of the 1TDM after Löwdin orthogonalisation

Ω_{AA} Weight of local excitations on fragment A

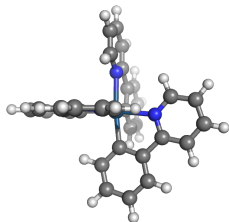
$\Omega_{AB}, A \neq B$ Amount of charge transfer from A to B

- **Fragment** definition
 - Central metal + ligands: Ir, L^1 , L^2 , L^3

¹FP, H. Lischka, *JCTC* **2012**, 8, 2777.

²FP, M. Wormit, A. Dreuw, *JCP* **2014**, 141, 024106.

Charge Transfer Numbers

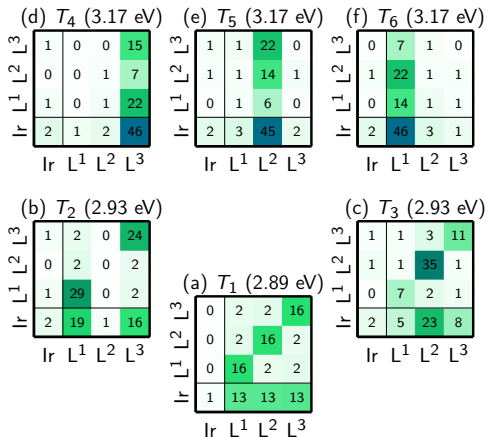


► $T_1 - T_3$

- $L^i \rightarrow L^i$ (IL)
- $\text{Ir} \rightarrow L^i$ (MLCT)

► $T_4 - T_6$

- $\text{Ir} \rightarrow L^i$ (MLCT)
- $L^i \rightarrow L^j$ (LLCT)



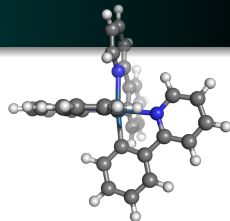
Charge Transfer Numbers

► General classification

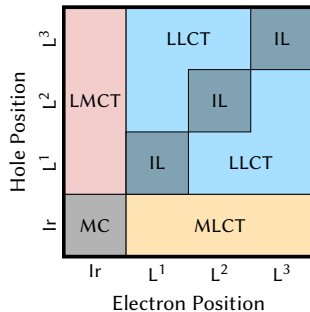
- Different formal state characters correspond to different Ω_{AB} elements

→ Automatic **classification** of state character

→ Quantification of **state mixing**



	E (eV)	Sym.	State character	
T_1	2.89	A	49% IL	38% MLCT
T_2	2.93	E	53% IL	35% MLCT
T_3	2.93	E	53% IL	35% MLCT
T_4	3.17	$A+E$	49% MLCT	31% LLCT
T_5	3.17	$A+E$	50% MLCT	31% LLCT
T_6	3.17	$A+E$	50% MLCT	32% LLCT

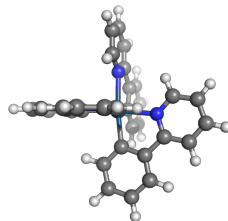


¹FP, A. Dreuw, *JPCA* **2015**, 119,1023.

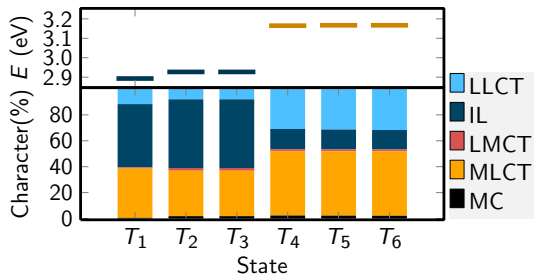
²S. Mai, FP, J. Dorn, M. Fumanal, C. Daniel, L. González, *CCR* **2018**, 361, 74.

Charge Transfer Numbers

	E (eV)	Sym.	State character	
T_1	2.89	A	49% IL	38% MLCT
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► Compact graphical depiction



❓ Can we avoid the fragment definition

Exciton Analysis

Exciton analysis

- Interpret the 1TDM as the wavefunction χ_{exc} of the electron-hole pair
- Use as a basis for analysis

Exciton wavefunction

$$\chi_{exc}(x_h, x_e) = \sum_{\mu\nu} D_{\mu\nu}^{0I} \chi_{\mu}(x_h) \chi_{\nu}(x_e)$$

$D_{\mu\nu}^{0I}$ Matrix representation of the 1TDM

χ_{μ} Atomic orbital

x_h, x_e Coordinates of the **excitation hole** and the **excited electron**

¹S. A. B  ppler, F.P. M. Wormit, A. Dreuw, *Phys. Rev. A* **2014**, 90, 052521.

Exciton Analysis

Operator expectation value

$$\langle \hat{O} \rangle = \frac{\langle \chi_{exc} | \hat{O} | \chi_{exc} \rangle}{\langle \chi_{exc} | \chi_{exc} \rangle}$$

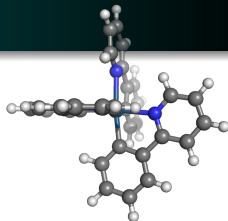
Exciton size

$$d_{exc}^2 = \langle (r_e - r_h)^2 \rangle$$

- ▶ Average separation of the electron and hole quasi-particles
 - Evaluated using **analytic integration** techniques
- ☺ No fragment definition
- ☺ No population analysis

¹S. A. Bäppler, FP, M. Wormit, A. Dreuw, *Phys. Rev. A* **2014**, 90, 052521.

Exciton size



► TDDFT/B3LYP

- Ir(ppy)₃

	E (eV)	State character		d_{exc} (Å)
2^3A (T_1)	2.74	49% IL	38% MLCT	4.07
1^3E (T_2, T_3)	2.77	51% IL	37% MLCT	3.94
3^3A (T_4)	2.97	49% MLCT	29% LLCT	4.40
2^3E (T_5, T_6)	2.98	48% MLCT	31% LLCT	4.50
3^3E (T_7, T_8)	3.10	48% MLCT	38% LLCT	5.03
4^3A (T_9)	3.14	47% MLCT	38% LLCT	5.05

► Smaller exciton size → less CT character

❓ What is really going on

Electron/hole densities

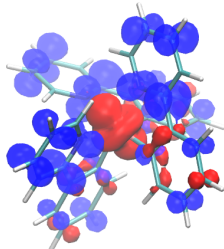
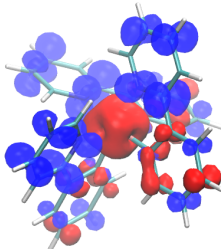
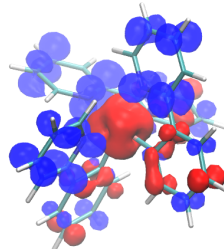
Density for the excited electron / excitation hole

$$\rho_e(r_e) = \int \gamma^{0I}(r_h, r_e)^2 dr_h$$

$$\rho_h(r_h) = \int \gamma^{0I}(r_h, r_e)^2 dr_e$$

- Equivalent to weighted sums over NTOs

Electron/hole densities

 2^3A  3^3A  4^3A 

- ▶ **Hole** density → centered around Ir
- ▶ **Electron** density → delocalized

❓ Can we learn more

Conditional densities

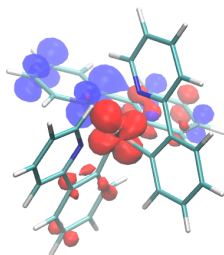
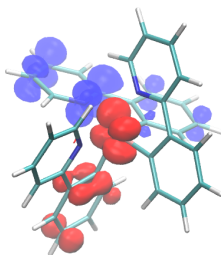
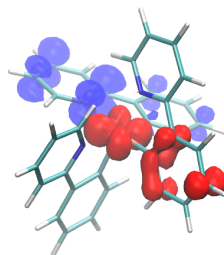
Conditional density for the excited electron

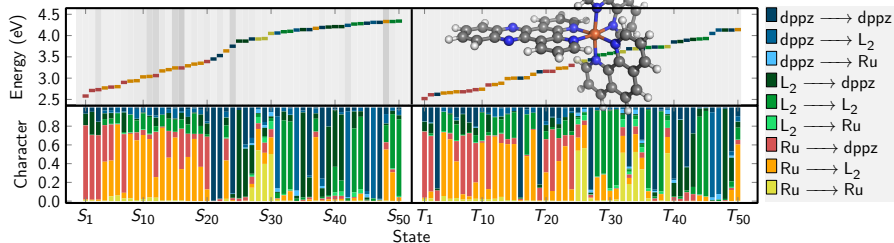
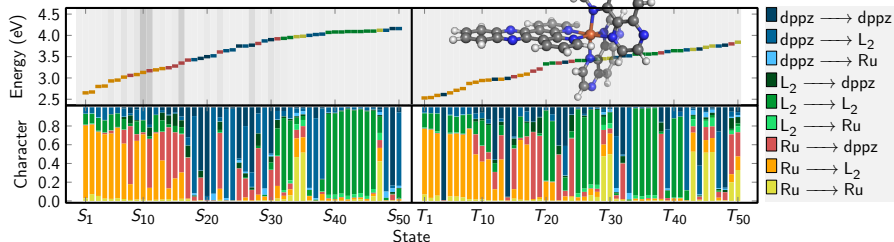
$$\rho_e^{h:A}(r_e) = \int_A \gamma^{0I}(r_h, r_e)^2 dr_h$$

$\rho_e^{h:A}(r_e)$ Conditional **electron** density for the **hole** localized on fragment A

- Evaluated through multiplication of matrix blocks

Conditional electron/hole densities

 2^3A *Local* 3^3A *"Corner" CT* 4^3A *"Straight" CT*

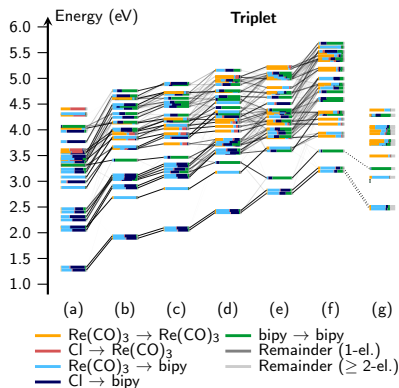
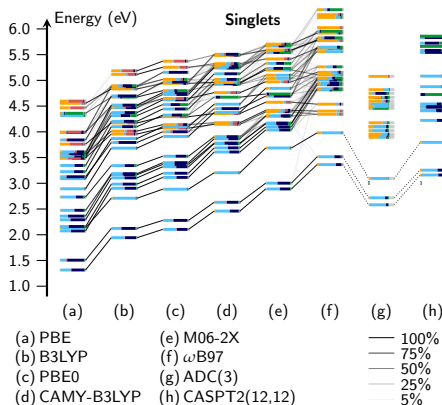
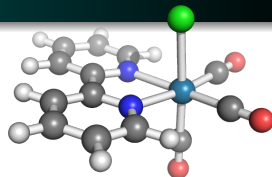
$[\text{Ru}(\text{phen})_2(\text{dppz})]^{2+}$  $[\text{Ru}(\text{tap})_2(\text{dppz})]^{2+}$ 

¹S. Mai, F.P. J. Dorn, M. Fumanal, C. Daniel, L. González, *CCR* **2018**, 361, 74.

Method evaluation

► Re(Cl)(CO)₃(bipy)

- Comparison of TDDFT and wavefunction based methods



¹S. Mai, FP, J. Dorn, M. Fumanal, C. Daniel, L. González, *CCR* **2018**, 361, 74.

Conclusions

- ▶ Excited state **wavefunction analysis** tools
- ▶ Visualization
 - Natural transition orbitals
 - *Conditional electron densities*
- ▶ Quantitative analysis
 - Charge transfer numbers
- Automatization
- Rigorous discussion
- ▶ Implementation through **matrix operations**
- Fast

Application areas

► Transition metal complexes

- FP, A. Dreuw, *JPCA* **2015**, 119, 1023.
- S. Mai, FP, J. Dorn, M. Fumanal, C. Daniel, L. González, *CCR* **2018**, 361, 74.
- *Talk*: Leticia González
- *Poster*: Pedro Sánchez-Murcia

► DNA

- J. J. Nogueira, FP, L. González, *Chem. Sci.* **2017**, 8, 5682.

► Conjugated polymers

- S. A. Mewes, J.-M. Mewes, A. Dreuw, FP, *PCCP* **2016**, 18, 2548.
- S. A. Mewes, FP, A. Dreuw, *JPCL* **2017**, 8, 1205.

► Push-pull systems

- P. Kautny, F. Glöcklhofer, T. Kader, J. Mewes, B. Stöger, J. Fröhlich, D. Lumpi, FP, *PCCP* **2017**, 19, 18055.

► Method evaluation

- FP, S. A. Mewes, A. Dreuw, L. González, *JCTC* **2017**, 13, 5343.
- S. A. Mewes, FP, A. Krylov, A. Dreuw, *JCTC* **2018**, 14, 710.

Software

Extended *wavefunction analysis toolbox*.

TheoDORE - **Theo**retical **D**ensity, **O**rbital **R**elaxation and **E**xciton analysis¹

- ▶ Program package for wavefunction analysis
- ▶ Interfaces to various quantum chemistry programs:
Columbus, Turbomole, Orca, GAMESS, Gaussian, ADF, Terachem,
...
- ▶ Open-source

libwfa - An open-source wavefunction analysis tool library²

- ▶ **Q-Chem**: ADC, EOM-CC, TDDFT
- ▶ **MOLCAS**: CASSCF, MS-CASPT2

¹<http://theodore-qc.sourceforge.net>

²<https://github.com/libwfa/libwfa>

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H. Lischka



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Alexander von Humboldt
Stiftung/Foundation



FWF

Slides available at: <https://fplasser.sci-public.lboro.ac.uk>